

Industrial Applications of Molecular Modeling

July 2000

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- Pharmaceuticals & Food Additives
 - Merck – Merck Molecular Force Field for Aqueous Biomolecules
 - Nutrasweet – Rheology of Thickening Agents
 - Unilever – Copolymer Composition of Emulsifiers
- Commodity Chemical
 - Dow – Accurate Thermodynamic Data for Hazard Analysis
 - DuPont – From Process Design to Environmental Fate
- Petrochemical
 - Amoco – Thermodynamics of Intermediates in Design of Catalytic Cycle
- Automotive
 - Lubrizol – Isolating Effect of Individual Components of Complex Product Mixtures, Catalyst Substitution
 - Ford – Improved 3-Way Catalyst Design for Fuel Economy
- Coatings & Colors
 - PPG – Electronic Transitions to Design Photochromics
 - Eastman Kodak – Colors and Electrostatics for Imaging Systems, Acylation Process Improvement

Merck – Thomas Halgren

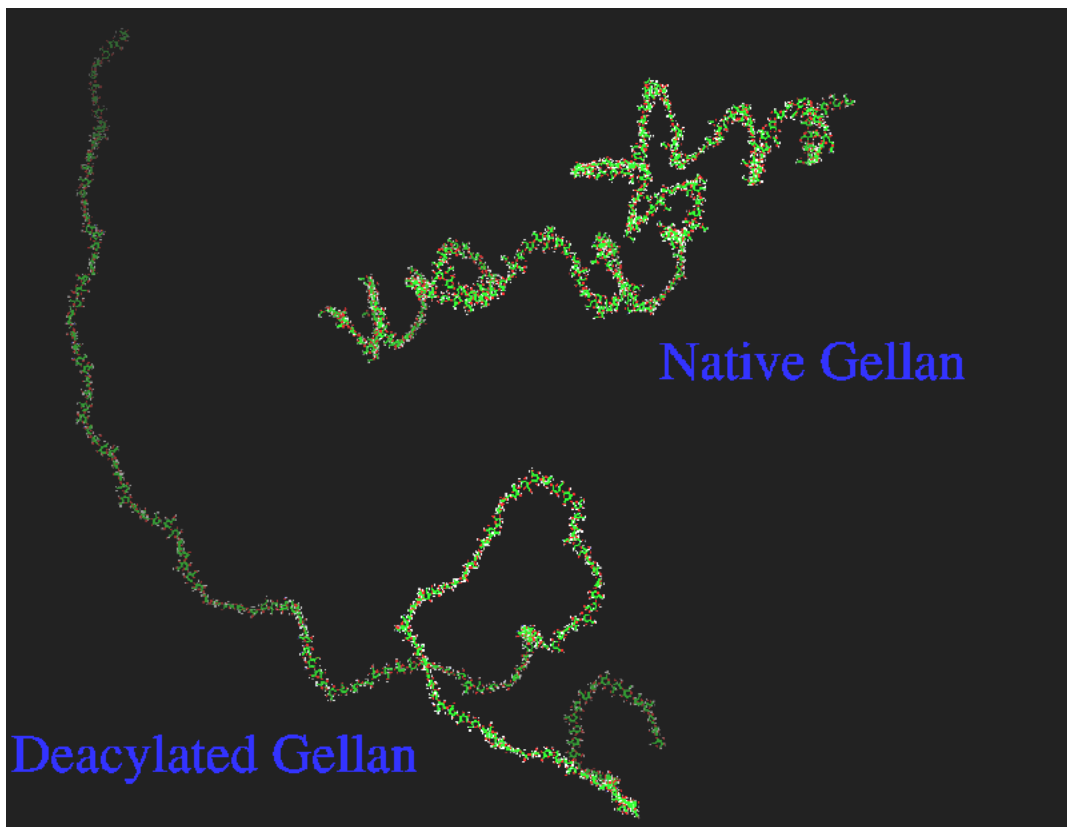
More accurate Molecular Mechanics specific for biomolecules in aqueous environment: drug docking & design, protein folding, CoMFA (Comparative Field Analysis).

- Merck Molecular Force Field (MMFF94)
 - Ab Initio (MP2/6-31G*) Derived Parameters
 - Class II Force Field
 - Quartic Term (correct for anharmonicity)
 - Cross Term (coupling of vibrations)
- Improved Energies
 - Better Conformational Barriers
 - Better Hydration of Ions (2.9 vs. 8.2 kcal w/ MM2)

NutraSweet Kelco – Todd Telashek

Control rheology of acylated polysaccharide gelation agents used as thickening agents for food and other applications.

- Model chain extension of chains with various levels of acylation by RIS-MC.

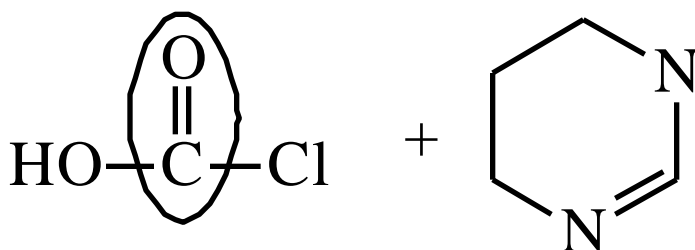


- Trade off brittleness with extended chains at low acylation levels vs elasticity at high acylation levels.

DOW – Nelson Rondan, David Frurip

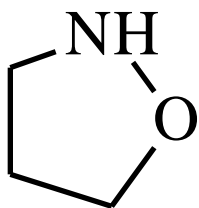
Indirect calculation of thermodynamics for hazard analysis and process design.

- Larger Molecules than DuPont
Benson Group Contribution Method in CHETAH
- High Level Ab Initio Calculations (G2) to Fill in Missing Groups



25 Hrs. for MP4/6-311G**

- Identified Explosion Hazard by Heat of Decomposition



80% of TNT

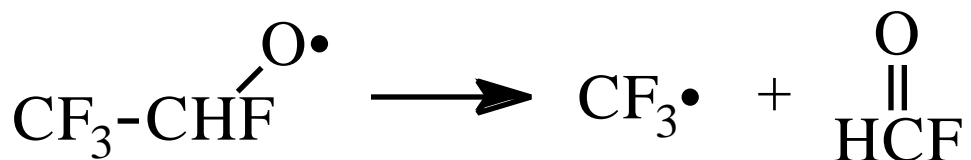
DuPont – Dave Dixon

Direct calculation of accurate thermodynamics and kinetics for complete product cycle.

- Grand Challenge – design chemical plant from scratch
 - Thermodynamics
 - Kinetics
 - Catalyst Design
 - Process Simulation
 - Fluid Dynamics
 - etc.

- CFC Alternatives – HFC-134A
 - Intermediates in Potential Process for CF₃CH₂F
 - NIST Measurement – \$50K + 90 Days
 - MP2/TZP Calculation – \$5k + 7 days

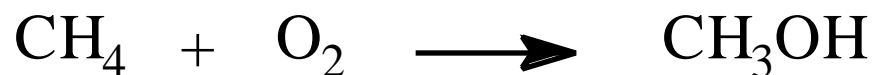
Predict Environmental Fate
Atmospheric Kinetics



Amoco – Joe Golab

Find thermodynamic balance in design of catalytic cycle.

- Upgrade Value of Natural Gas: CH₄



1st Step in Oxidative Coupling to Higher Hydrocarbons

- Ta Oxides React with CH₄

Alkylates Oxide

Doesn't Release Methanol

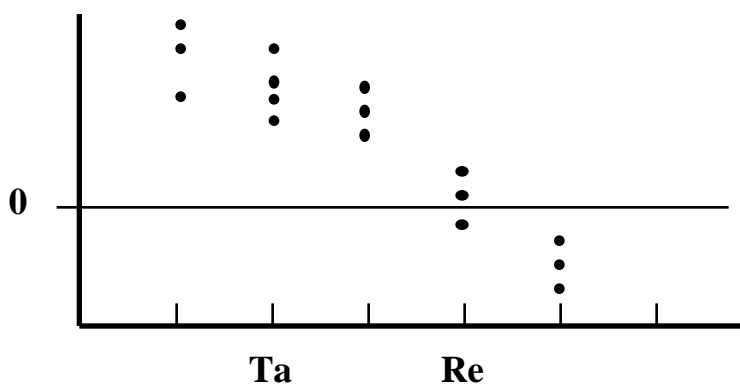
- Model 3rd Row Transition Metal Oxide Families

Oxidation, Coordination, and Spin States

ECP's for Relativistic Atoms

Alkylation Thermodynamics

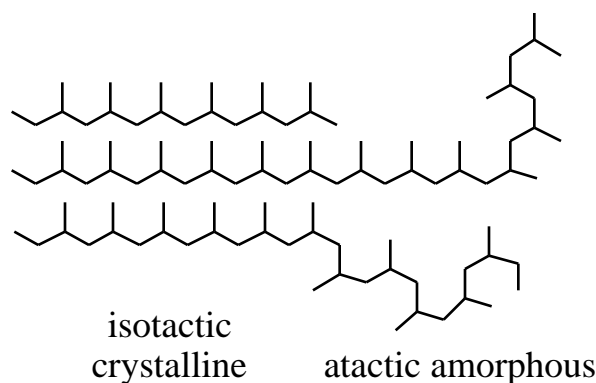
Barrier to Methanol Dissociation



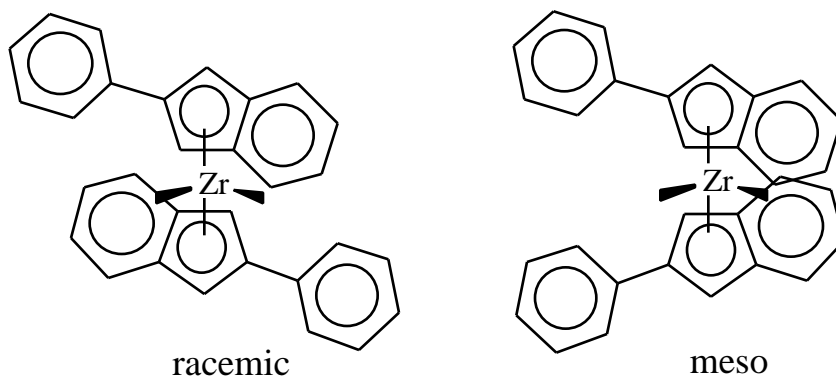
Petrochemical Companies – Bob Weymouth, Others

Design of new generation olefin polymerization catalysts: elastomeric PP by control of tacticity in blocks.

- Physically Cross-linked Blocks



- Model Multiple Forms of Single Site Catalyst
Relative Energies by DFT for Transition Metal



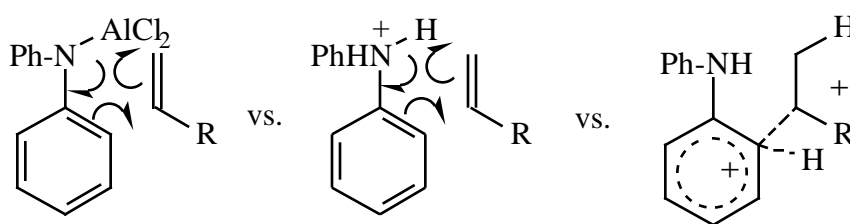
Relative Run Lengths vs. Rate of Change in Catalyst Form
Stereochemical Control vs. Termination (MW)

Lubrizol – Anne Chaka

Study intermediates and components of lubricant additives not otherwise isolable: reduce waste, design continuous process, optimize additive.

- Test Cheaper/Safer Replacements for AlCl_3 & BF_3 in Additive Synthesis

Transition State Modeling



Can't combine catalysts in continuous process!

- Optimum Polysulfide for Corrosion Inhibition

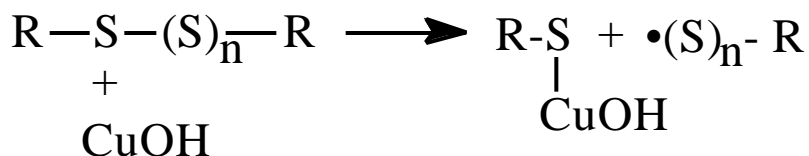
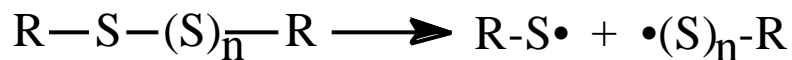


$n = 1-5$ protects Fe in gears

$n > 2$ corrodes Cu alloys

DFT Fast Structure Code

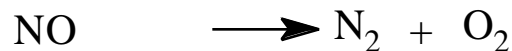
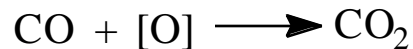
Model Stability w/ & w/o Cu Oxide "Catalysis"



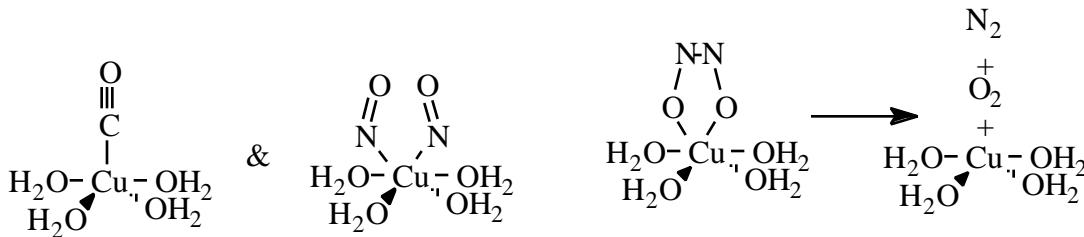
Ford - Bill Schneider

Design exhaust catalyst for pollution control with lean operation (excess O₂) for increased fuel economy. Manipulate Cu environment in zeolite to avoid NO_x in 3-Way Catalyst. Control BaO surface to store NO_x w/o BaSO₄ in Dual Mode Converter.

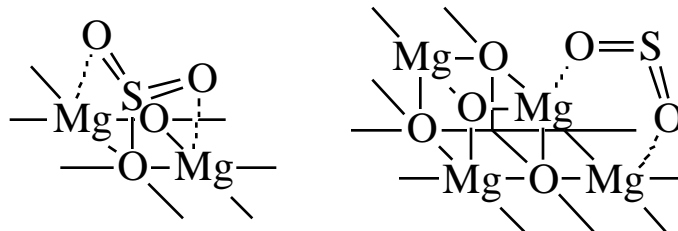
- Catalysts Convert CO and NO_x



- Model Binding of CO, NO, H₂O to Supported Cu/ZSM-5 by DFT.



- Model binding of NO_x, SO₂ & SO₃ on Ba/MgO [100] surface and step with Car-Parinello DFT/MD.



Procter & Gamble – Bill Laiding

Understand unique property of component in support of deodorant patent.

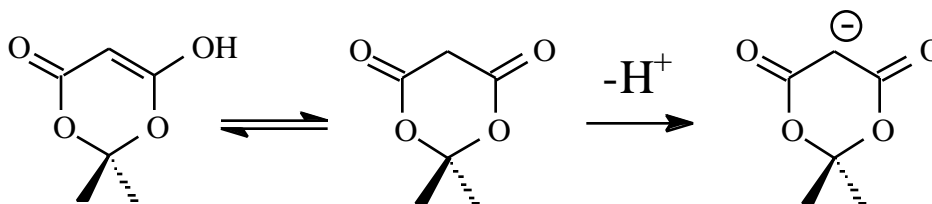
- Meldrum's Acid Unusually Acidic

$\text{pK}_a(\text{DMSO}) = 7.3$

Acyclic (Malonate) $\text{pK}_a(\text{DMSO}) = 15.9$

- Model Molecules, Tautomers, and Ions to Calculate Acidity

MP2/6-311⁺⁺G^{**} for ions



Enol Disfavored

Different Acidities for Axial & Equatorial H

PPG – Rick Ross

Design inorganic pigments & predict photochromic colors and activity.

- **Color an Electronic Process**

Promote an Electron into an Unfilled Orbital

Energy Difference Determines Wavelength Absorbed

ZINDO CI for Good Unfilled Orbital Energies

- **Photochromism**

Color vs. MeO- substitution on PNA's

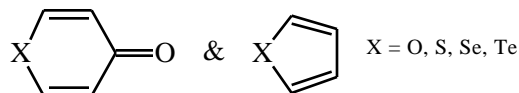
Promote Electron from Excited State Instead of Ground State

Eastman Kodak – John McKelvey, Peter Margl

Checking experimental results, predicting electronics and colors for film and photocopier application, alkylation process improvement.

- Chalcogenic Heterocycles

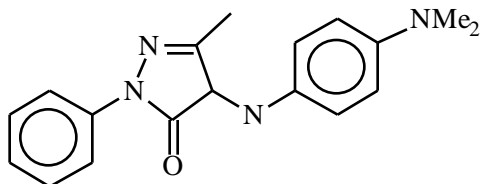
Stabilization of Charge Important for Photocopier
Anomalous Experimental Se Pyran Dipole Moment



Range of Methods, PM3 Through DFT
Identified Error in Workup of Lab Data!

- Predicting Color of Pyrazalene Dyes

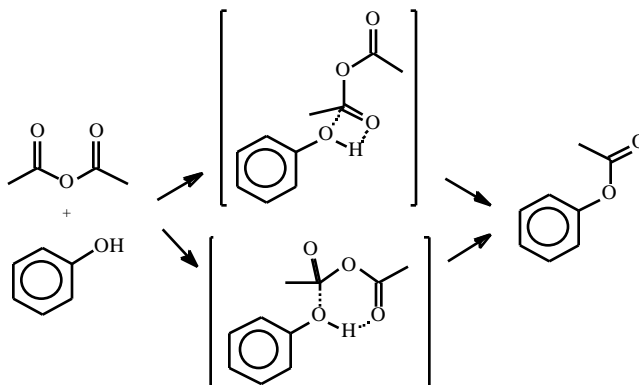
ZINDO for Large Systems



Color & Intensity Controlled by Geometry

- Accelerate Phenol Acylation Process.

Acid vs Base Catalysis of Acyl Transfer Reaction
B3LYP with SCI-PCM for Solvation.



Acyl Transfer Rate Not Limiting; Solubility Limited!